

Hydrogen separation using membranes

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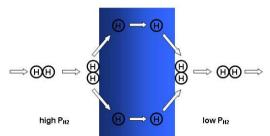


Scope

- High-temperature H₂-selective membranes
- The role of membranes in pre-CC
- Catalytic membrane reactors
- CMR performance characteristics
- CMR optimisation



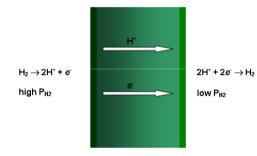
High-temperature H₂-selective membranes



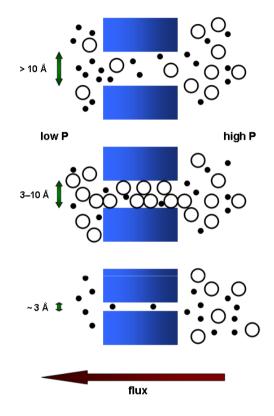
Alloy (Pd, Pd-coated V)

⇒ ⊕⊕ ⇒ 300-600C

Pure H2 product



Dense ceramic 600C Pure H2 product



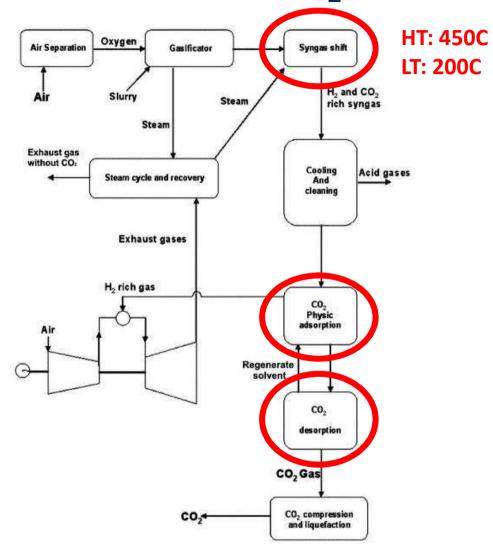
Microporous (eg, SiO2)

< 300C

< 100% H2 (depending on pore size)



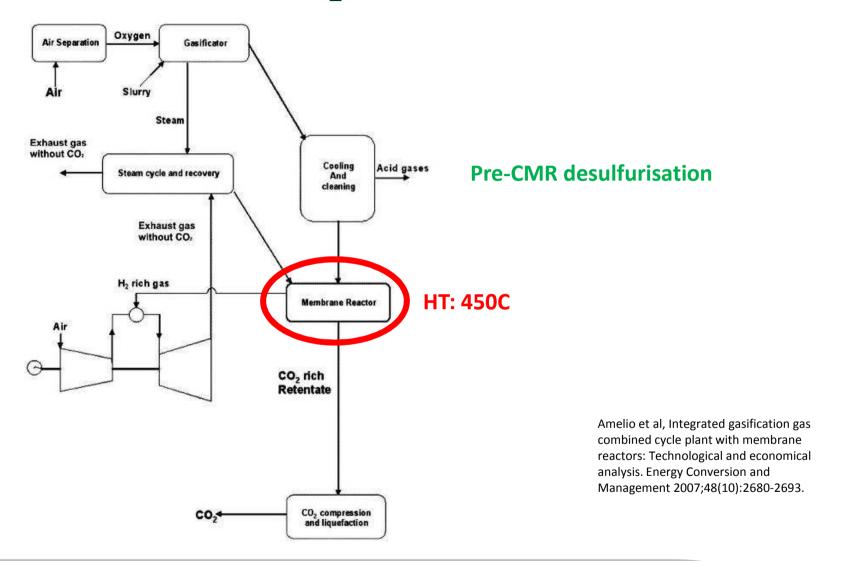
Pre-combustion CO₂ capture (conventional)



Amelio et al, Integrated gasification gas combined cycle plant with membrane reactors: Technological and economical analysis. Energy Conversion and Management 2007;48(10):2680-2693.

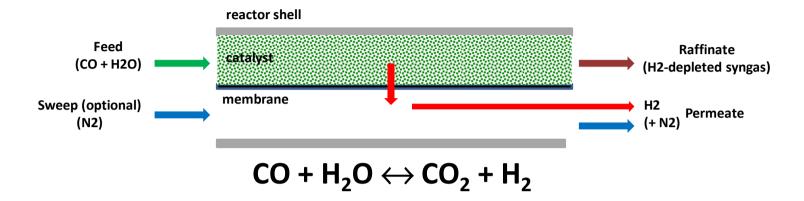


Pre-combustion CO₂ capture (membrane reactor)





The catalytic membrane reactor



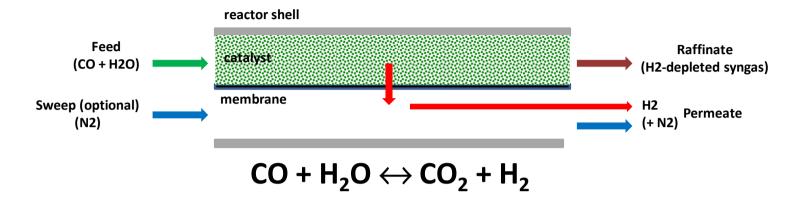


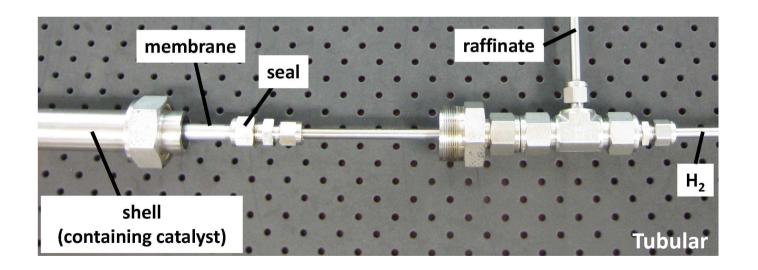


Planar



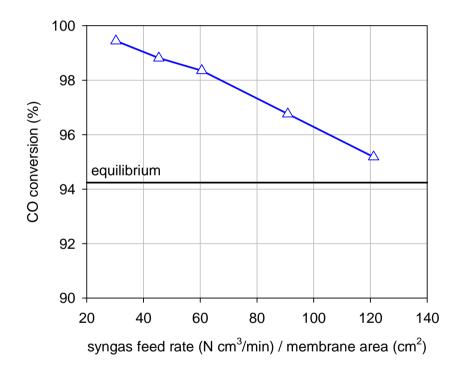
The catalytic membrane reactor







CMR performance characteristics



Removing H2 from reactor promotes forward WGS reaction

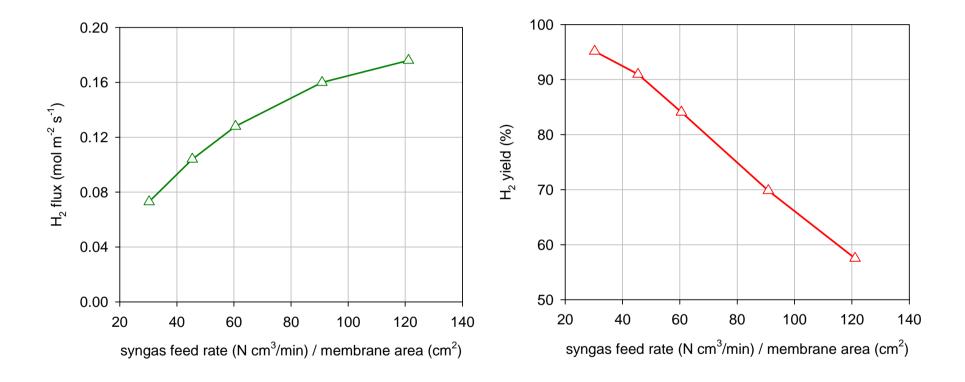
Creates artificially-high equilibrium

Eliminates requirement for low-temperature WGS reactor

400C, 20 bar, 3:1 H₂O:C



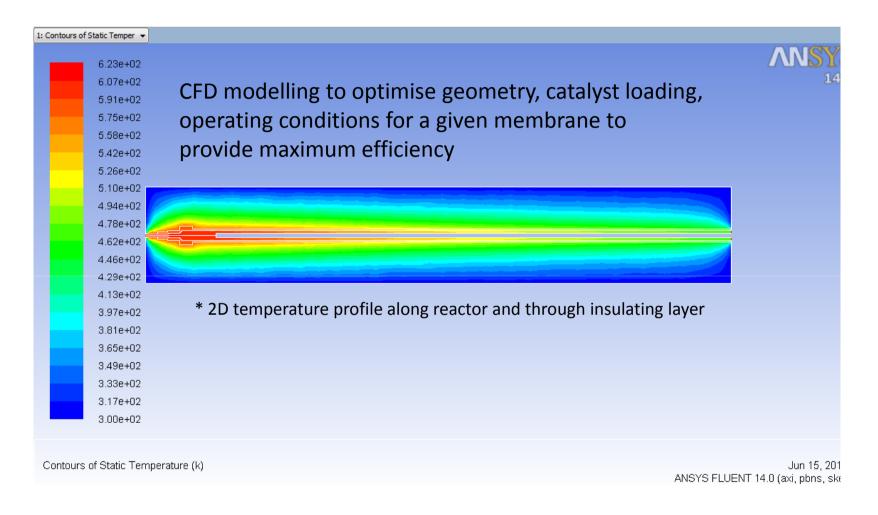
CMR performance characteristics



400C, 20 bar, 3:1 H₂O:C

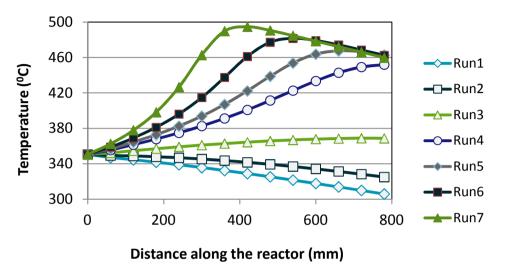


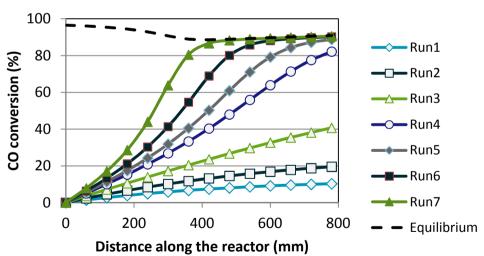
Modelling heat flow and reaction rate





Modelling heat flow and reaction rate





Temperature varies along reactor length due to i) exothermic WGS reaction and ii) radiative and conductive heat losses

Membranes operate in fairly narrow temperature ranges (350-450C for V-based alloy membranes)

Catalyst loading and feed flow rates must be tailored to minimise temperature gradient along reactor length



Summary

- Membranes: alloy membranes are infinitely selective to H2
 - Can be used as a stand-alone H2/CO2 separator to produce pure H2, or in a water-gas-shift membrane reactor
- Materials issues: the membrane is the key component
 - Must provide high H2 flux, low cost, H2S tolerance, tolerance to thermal cycling
- Catalytic Membrane Reactor: offers process intensification by combining several shift and separation stages in a single reactor
 - CO2 captured pre-combustion at high pressure; chemical energy in syngas shifted from CO to H2 for use in turbine, fuel cell, chemical synthesis, etc.



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